

## Engel 9.13

**Problem.** Calculate the mean value of the radius  $\langle r \rangle$  at which you would find the electron in a hydrogen atom in the 210 state.

**Answer.** The 210 ( $2p$ ) wave function is given on Engel p. 164. The mean value of  $r$  is the expectation value of  $r$ .

Integration must be performed with respect to all three spherical polar coordinates, but the radial integral will contain an extra factor of  $r$  because we are trying to establish the expectation value of this quantity.

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Clear[ψ210, N210, R210, θ210, a0]
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$$N_{210} = \frac{1}{4\sqrt{2}\pi} \left(\frac{1}{a_0}\right)^{3/2}$$

$$R_{210} = \left(\frac{r}{a_0}\right) e^{-r/(2a_0)}$$

$$\theta_{210} = \cos[\theta]$$

$$\psi_{210} = N_{210} R_{210} \theta_{210}$$

$$\frac{\left(\frac{1}{a_0}\right)^{3/2}}{4\sqrt{2}\pi}$$

$$\frac{e^{-\frac{r}{2a_0}} r}{a_0}$$

$$\cos[\theta]$$

$$\frac{\left(\frac{1}{a_0}\right)^{5/2} e^{-\frac{r}{2a_0}} r \cos[\theta]}{4\sqrt{2}\pi}$$

The wave function is independent of  $\phi$ , so we will do this integration first

$$\phi_{\text{int}} = \int_0^{2\pi} d\phi$$

$$2\pi$$

The wave function contains a  $\theta$ -dependent term, and we must remember to use  $\sin\theta d\theta$  as the index of integration

$$\theta_{\text{int}} = \int_0^\pi \theta_{210}^2 \sin[\theta] d\theta$$

$$\frac{2}{3}$$

The integration with respect to  $r$  must also include an extra factor of  $r$  because this is the operator whose expectation value we want to determine, and  $r^2 dr$  as the index of integration

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Clear[Rint]
Rint =  $\int_0^{\infty} R_{210}^2(r) (r^2) dr$ 
Rint = Simplify[Rint, a0 > 0]
 $\frac{1}{a_0^2}$  If[Re[a0] > 0, 120 a06, Integrate[e- $\frac{r}{a_0}$  r5, {r, 0,  $\infty$ }, Assumptions  $\rightarrow$  Re[a0]  $\leq$  0]]
120 a04

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$a_0$  is the Bohr radius and is a positive real quantity, so this fact can be used to obtain the necessary integral. Now we combine the normalization constant and the three integrals ...

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N2102  $\phi_{int}$   $\theta_{int}$  Rint
5 a0

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The mean radius of the electron is 5 Bohr radii, about 2.5 angstroms, for this energy state.